NAME	
DATE	PERIOD

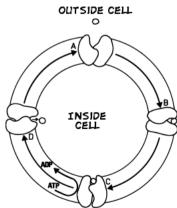
Cell Membrane & Tonicity Worksheet

Composition of the Cell Membrane & Functions		
The cell membrane is also called the m	embrane and is made of a	SKETCH AND LABEL a
phospholipid The phospholipids have a	hydrophilic (water attracting)	phospholipid coloring
and two hydrophobic (water repelling)	The head of a	the heads red and the tails blue.
phospholipid is made of an alcohol and grou	p, while the tails are chains of	tans side.
Phospholipids can move	and allow water and other	
molecules to pass through into or out of the o	cell. This is known as simple	
because it does not require and	nd the water or molecules are	
moving the concentration gradient.		
Another type of lipid in the call membrane is	that makes the membrane me	co fluid
Another type of lipid in the cell membrane is the Embedded in the phospholipid bilayer are the tell membrane is the phospholipid bilayer are the tell membrane is the phospholipid bilayer are	that makes the membrane more	ognition
Proteins called proteins go all the way th		
only on one side. Integral proteins are also called		
carbohydrates use proteins to help move across cell men		
carbohydrate attached to help cells in recog	nize each other and certain molecule	25.
List 4 functions of the cell or plasma membrane:		
a		
b		
C		
d		
Correctly color code and identify the name for each part of Letter Phospholipid bilayer (no color)		
and the second s	d	P .
Match the cell membrane structure or its function with the		•
Letter Structure/Function	Letter Structure	e/Function
Attracts water	Repels water	
Helps maintain flexibility of membrane	Make up the bilayer	
Involved in cell-to-cell recognition	Help transport certain	materials across the
	cell membrane	

Define osmosis In which direction does	water move acros	s membranes, up or do	own the conc	entration gradie	nt?
Define these 3 terms: a. isotonic- b. hypertonic c. hypotonic			 		
Use arrows to show the isotonic environment liggreen.	ght blue, the hypot				
	98% WATER	98% WATER	98 WAT	D	
	95% WATER	98% WATER	100% V	NATER	
Match the description o	r picture with the o	osmotic condition:			
A. Isotonic				e concentration	
B. Hypertonic		solution in wh		concentration	is the same
b. Hypertonic		condition that	-		
C. Hypotonic		red blood cell plant cell loses	bursts (cytoly s turgor press a higher solut good turgor	ysis) sure (Plasmolysi se concentration pressure	•
Label the tonicity for ea	ch solution (isoton	ic, hypotonic, or hyper	tonic):		
H ₂ O H ₂ C	H ₂ O	H ₂ O		H ₂ O H ₂ O	H ₂ O
					
Transport Requiring Ene	ergy				
What type of transport is	s represented by t	he	(OUTSIDE CELL	
following picture?				\sim	

What type of transport is represented by the following picture? _____ What energy is being used? _____ In which direction (concentration gradient), is the movement occurring? _____

Color the internal environment of the cell yellow. **Color and Label** the transport proteins red and the substance being moved blue.



One type of active transport is called the pump which helps muscle cells	ATP ADP	
contract. This pump uses to move ior	ns 🖺 😨	
the concentration gradient. The prote	ein The Total	membrane
that is used to pump the ions through is called a protein and it changes its		mem
to move the ions across the cell membrane.		
Label and color the carrier proteins red and the ions		
green.		
	②	
TONICITY AND OSMOSIS		_
_		
key:	plant c	رك
solute particle	I the state of	, ,
cell membrane		?
in all solutions, the solvent is H ₂ O	B C animal	cell
Post I Fill in the blanks		
Part I – Fill in the blanks. A is a fluid in which a s A is a substance dissol	substance is dissolved.	
A is a substance dissol	lved in a solvent.	
A is a combination of s The process by which water diffuses across a mem	Solute and Solvent.	
The process by which water diffuses across a mem		
Part II – Look at the solutions illustrated above an	nd fill in the blanks.	
1. Solution B is to Solut	tion A. This is because Solution B has a greater	
concentration of in it than o	does Solution A. Solution C has no solutes dissolved in i	t,
therefore it is to both So	olutions A and B.	
2. As a relative concentration of solutes in two solu	utions increases, of necessity the concentration of wate	r in
	tion A has a lower concentration of tha	
does Solution C ; Solution A is also hypertonic to S		••
does solution e, solution A is also hypercome to s	oriation C.	
3. If you wanted to make Solution A isotonic to So	Dolution B , you could add water to Solution or you	
could add solute to Solution If you took all the	hree solutions, put them into a large container and mixe	èd
them thoroughly, then redistributed the solution a	among three containers, Solution A would be	to
Solution B. Solution A would also be	to Solution C, and Solution C would beto	0
Solution B.		
Part III – Look at the solutions and cells illustrated	d above and fill in the blanks	
		21/
, , ,	imal cell have equal concentrations of solutes, we can sa	ıy
their cytopiasms are to each other	r. If we put both the plant and the animal cells into	

Solution A, we would expect no change in the ce	lls, because Solutio r	n A is t	o the cytoplasm of
each cell.			
2. Let's put both cells into Solution B. Because So	olution B is hyperton	ic to the cytoplasm	s of the cells, we
would expect water to the cells thro	ugh the process of _	This	would result in the
cytoplasm of both cells shrinking.			
3. Now we'll put both the plant and animal cell in	to Solution C, which	, because it contains	s no solutes at all, is
to the cytoplasm of both	cells.	will enter bo	th cells through
osmosis. The animal cell is likely to	, unfortunately. 1	The plant cell , howe	ver, is protected
from this because of the presence of its	·		
Before osmosis After osmosis	Cell in Isotonic Solution	Cell in Hypotonic Solution	Cell in Hypertonic Solution
	Water molecules Dissolved particles	00000	
· Water molecule			
• Sugar molecule			
	H ₂ O	H ₂ O	H ₂ O
Selectively	In an isotonic solution, water molecules move into and out of	In a hypotonic solution, water enters a cell by osmosis, causing	In a hypertonic solution, water leaves a cell by osmosis, causing
permeable ————————————————————————————————————	the cell at the same rate.	the cell to swell.	the cell to shrink.

Refer to the U-tube pictures above when answering the questions below.

- 1. Why did the number of water molecules on each side of the membrane change, whereas the number of sugar molecules stayed the same?
- 2. How does the plasma membrane of a cell compare with the membrane in the U-shaped tube?
- 3. Explain the behavior of water molecules in the isotonic solution.
- 4. Does osmosis occur if a cell is placed in an isotonic solution?
- 5. Why does water enter a cell that is placed in a hypotonic solution?
- 6. What happens to the pressure inside a cell that is placed in a hypertonic solution?
- 7. What can happen to animal cells when placed in a hypotonic solution? Explain.
- 8. What causes a plant to wilt?